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This book is ideal for use in a one-semester introductory course in physical chemistry for students of life sciences. The author's aim is to emphasize the understanding of physical concepts rather than focus on precise mathematical development or on actual experimental details. Subsequently, only basic skills of differential and integral calculus are required for understanding the equations. The end-of-chapter problems have both physiochemical and biological applications.

The Student Solutions Manual is written by Raymond Chang and Ken Goldsby. This supplement contains detailed solutions and explanations for even-numbered problems in the main text. The manual also includes a detailed discussion of different types of problems and approaches to solving chemical problems and tutorial solutions for many of the end-of-chapter problems in the text, along with strategies for solving them. Note that solutions to the problems listed under Interpreting, Modeling & Estimating are not provided in the manual.

Nothing can better help students understand difficult concepts than working through and solving problems. By providing a strong pedagogical framework for self study, this Solutions Manual will give students fresh insights into concepts and principles that may elude them in the lecture hall. It features detailed solutions to each of the even-numbered problems from Raymond Chang and Jay Thoman's Physical Chemistry for the Chemical Sciences. The authors approach each solution with the same conversational style that they use in their classrooms, as they teach students problem solving techniques rather than simply handing out answers. Illustrative figures and diagrams are used throughout.

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Following in the wake of Chang's two other best-selling physical chemistry textbooks (Physical Chemistry for the Chemical and Biological Sciences and Physical Chemistry for the Biosciences), this new title introduces laser spectroscopist Jay Thoman (Williams College) as co-author. This comprehensive new text has been extensively revised both in level and scope. Targeted to a mainstream physical chemistry course, this text features extensively revised chapters on quantum mechanics and spectroscopy, many new chapter-ending problems, and updated references, while biological topics have been largely relegated to the previous two textbooks. Other topics added include the law of corresponding states, the Joule-Thomson effect, the meaning of entropy, multiple equilibria and coupled reactions, and chemiluminescence and bioluminescence. One way to gauge the level of this new text is that students who have used it will be well prepared for their GRE exams in the subject. Careful pedagogy and clear writing throughout combine to make this an excellent choice for your physical chemistry course.

Hailed by advance reviewers as "a kinder, gentler P. Chem. text," this book meets the needs of an introductory course on physical chemistry, and is an ideal choice for courses geared toward pre-medical and life sciences students. Physical Chemistry for the Chemical and Biological Sciences offers a wealth of applications to biological problems, numerous worked examples and around 1000 chapter-end problems.

The Student Solutions Manual will have all the solutions to the even numbered problems in the text. The style of the solutions will match worked examples in the text to help the student learn how to solve the problems.

Contains complete worked-out solutions for all "B" exercises and half of the end-of-chapter problems.

In this third edition, core applications have been added along with more recent developments in the theories of chemical reaction kinetics and molecular quantum mechanics, as well as in the experimental study of extremely rapid chemical reactions.
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